

Process Safety: A Review on Understanding a Safety Data Sheet (SDS)

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Abstract:

Safety Data Sheets (SDS), formerly known as Material Safety Data Sheets (MSDS) in the United States, contain important information for the safe handling of chemicals. This article discusses the use of safety data sheets in process design and safe chemical handling. The three main categories listed on the SDS (flammability, physical, and toxicity) are explained. Within each category, examples are provided for using specific information to design safe chemical processes. In addition, in the United States, the Occupational Safety and Health Administration (OSHA) revised its hazard communication standards in March 2012. In fact, this standard has been integrated into the UN's Globally Harmonized System of Classification and Labeling of Chemicals (GHS). The changes include requirements to use standard signal words, pictures, hazard statements, and warning statements.

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1.0 INTRODUCTION :

The use of chemical products to enhance and improve life is a common practice throughout the world. However, in addition to the benefits of these products, there may be negative effects on people or the environment. Therefore, many countries and organizations have passed laws or regulations over the years that require uniform information and provide information to chemical users through labeling and safety data sheets (SDS). Given the large number of chemical products available, no single entity can handle it alone. Providing information allows chemical users to understand the nature and hazards of the chemical and to provide appropriate protective measures to be implemented in the local use environment.

Given the widespread nature of the global chemical trade and the need for national initiatives to ensure the safe use, transport and disposal of chemicals, it was recognized that a global approach to classification and labeling was the basis for such an initiative.

When countries have the same information about the chemicals they import or produce locally, they can create structures to manage chemical exposure and protect people and the environment.

2. EXAMPLE: A SAFETY DATA SHEET FOR HEXANE

Recently, I asked some friends and graduate students to provide me with a safety data sheet for the chemical hexane in my native language. Interestingly, some of the labels on the safety data sheets are very similar around the world. Open the information on the safety data sheet in American English, Mandarin, Korean, and Farsi. The ultimate goal of the GHS is to achieve global harmonization of these tables in all languages. The attached example has 16 key sections. Under the GHS, all safety data sheets in all languages of the world will eventually be the same in 16 main sections. Although all sections contain important information, this article focuses on the sections that are useful for risk analysis. These sections are: Section 5, Fire Fighting Operations; Section 9, Physical and Chemical Properties; and Section 11, Toxicological Information.

3.0 EXAMPLES OF USES OF SAFETY DATA SHEETS IN DESIGN AND PRACTICE

Below are some of the key steps I took in the initial hazard analysis related to the solvent release room requirements and design.

3.1 THE MISSION

Mission: Ability to extract and store hexane in a safe manner. When needed, hexane can be safely distributed to other areas of the house at a rate of 4 liters per minute (upto 30 minutes).

3.2 SECTION 2 : HAZARD IDENTIFICATION

A reference to the Safety Data Sheet shown in Section 2 of the Appendix immediately informs about danger. Highly flammable liquid and flammable vapors. The fog causes lightning. May cause respiratory, eye and skin irritation. Organ damage may occur in animals.

3.3 SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES

Section 9 provides important information that a safety engineer can use in their hazard analysis. In this case, the molecular weight (86.2 g/mol) is given and the molecular formula is given. The boiling point is 68.9°C, the freezing point is -139.4°C. All of these help to understand the dangers of handling hexane. For example, there is little concern that hexane will freeze at room temperature. Hexane has a boiling point of 68.9°C. This is a problem. Movement in the pipe must be free from friction and heat sources to prevent the liquid from heating beyond its boiling point. If the pressure value of the pipe is too low to carry hexane, the temperature will exceed 68.9°C, and the pipe will burst due to high pressure due to evaporation. Therefore, a pressure relief system must be added to the storage tank and delivery line.

Adding a proper pressure relief system can reduce the risk of an explosion. Another parameter that is important for risk analysis in this case is the specific gravity of hexane water relative to water. Hexane has a water specific gravity of 0.66. Also, hexane cannot mix with water. If there is a leak, the liquid hexane will reach its lowest point, and if the floor drain of the room enters that low point, the hexane will flow into the house water tank. Since hexane is lighter than water, it dissolves in layers of water. This introduced flammable hexane into the building's water supply, causing a powerful explosion that destroyed the building. Therefore, the chamber design should include a second holding tank below the hexane-containing part of the process.

Vapor density parameter 3.0 relative to air. Hexane is three times denser than air. This number is very helpful in understanding negative situations. If the hexane escapes into the surro

ounding area, it sinks first. It is heavier than air. This means that the proper place for a smoke detector is at the bottom of the room, not at the top. Of course, it's in the specific parts of the LFL that Section 5 covers.

Another use of vapor density is to estimate the amount of water that, if spilled, would form a flammable mixture. Start with a room that is 6 meters long, 6 meters wide, and 3 meters high (108 cubic meters). The flammable limit of hexane is 1.7%. If 108 cubic meters x 1.7%, or 1.8 cubic meters, of hexane vapor enters the room, the room is likely to be flammable. How much hexane is this? At a temperature of 20°C, the density of air is 1.2 kg/cubic meter. Multiplying this value by three gives a density of hexane vapor of 3.6 kg/m³. Thus, if 3.6 kg/m³ times 1.8 m³ or 6.5 kg of hexane enters a room, a fire or explosion may occur. Using the specific gravity of water (0.66), this amount is equivalent to 10 liters of hexane spilled.

3.4 SECTION 11: TOXICOLOGICAL INFORMATION

Section 11 presents toxicological information for the chemical. Two main parameters are provided. These are LD50 and LC50 for species which are typically rats. LD50 provides estimates as to the dose if hexane is ingested that would kill 50% of the population at the dose rate provided. For example, hexane has a LD50 of 25 g/kg for rats (dose per mass). If 100 mice, and 0.25 kg (1/4 kg), each mouse ingests 7.5 g hexane (1/4 of the LD50 per kg) at one time, 50 mice will die in the few days. Also, exposure to hexane vapor is also performed. In this case, the parameter is called LC50 (lethal concentration that kills 50% of the population). If 100 mice breathe air containing 96,000 ppm (or 9.6% hexane) for one hour, 50 mice will die. The last parameter in this category is IDLH, Risk to Life or Health. It is 1,100 parts per million. This parameter is appropriate for workers who may be in environments containing hexane vapors. If the concentration of this environment exceeds 1,100 ppm, these workers risk their lives or health. In this case, the problems are related to fire and explosion. 1,100 ppm corresponds to the flammable limit and is about 1/10 of the flammable limit of 1.7%.



Fig 1 A universal transportation placard to be placed on trucks and drums containing flammable liquid such as hexane (sign background color is red)

4.0 POLICY AND REGULATORY FRAMEWORKS

Policy and regulatory frameworks play a crucial role in climate change resilience strategies. These frameworks are designed to guide both mitigation efforts, aimed at reducing the causes of climate change, and adaptation strategies, which prepare communities for the effects of a changing climate (“The Forest Policy Process, Regulatory, and Institutional Frameworks in Ethiopia”). They encompass a range of measures, including setting emissions reduction targets, enforcing environmental regulations, and promoting sustainable development practices. Importantly, these policies also focus on funding and investment in green technologies and infrastructure that can withstand climate impacts. Additionally, they often include provisions for research and development, aimed at advancing our understanding of climate change and developing innovative solutions (Waiti and Lorrenij). Education and public awareness campaigns are also key components, helping to cultivate a societal shift towards more sustainable practices. Collaboration



across different sectors and levels of government is

essential for the effective implementation of these policies, ensuring a coordinated and comprehensive response to the challenges posed by climate change. This multifaceted approach, underpinned by strong policy and regulatory frameworks, is essential to building resilient communities and safeguarding our environment for future generations (Waiti and Lorrenij).

Fig. 2 GHS pictogram symbols for n-hexane. See text above for the meaning of each one.

5. CONCLUSION

This document provides an example of using a Safety Data Sheet (SDS) for an initial hazard analysis of chemical use and storage, which contains three main sections: flammability, physical properties, and toxicology.

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